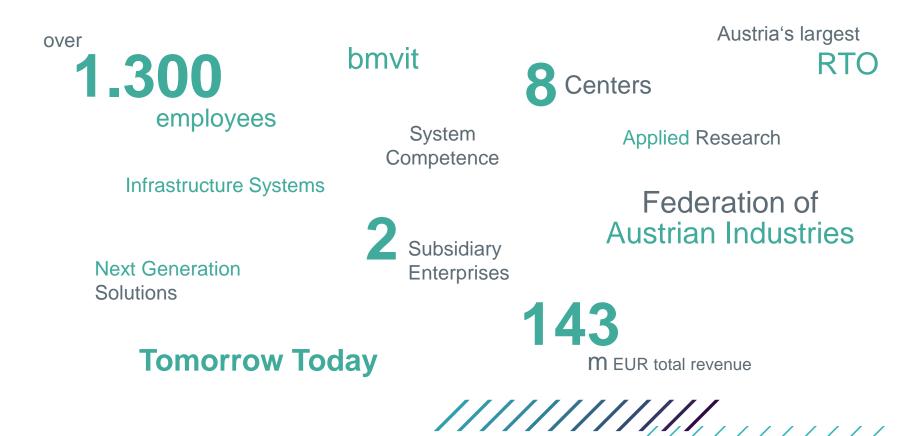


# **BATTERY INNOVATION**

#### From test tube to industrial production

Dr. Marcus Jahn November 12, 2018

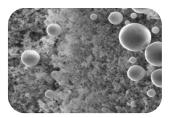






### CENTER FOR LOW-EMISSION TRANSPORT

#### **Research Fields**



Advanced battery materials



Battery processing technology and testing



Vehicle efficiency



Power electronics



Casting processes for high performance materials



Advanced forming processes and components



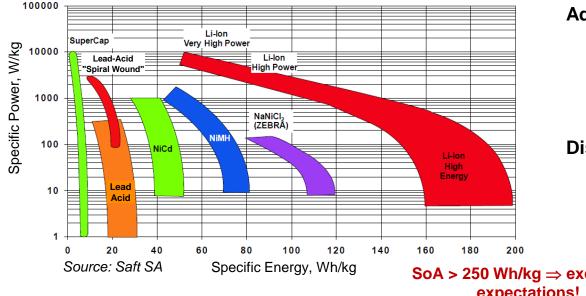
#### LITHIUM ION - MOTIVATION

- Energy storage Why do we need better systems?
- $\rightarrow$  \$150bn Market up to 2025





### WHY LITHIUM-ION TECHNOLOGIES?



#### **Advantages**

- High energy density
- Low self-discharge rate<sup>1</sup> •
- High cycling stability
- No "memory effect"<sup>2</sup>

#### Disadvantages

- Operating temperature: max. 60°C
- Sensitivity towards overcharge or deep discharge

SoA > 250 Wh/kg  $\Rightarrow$  exceeding expectations!

- LIBs dominate the markets for portable devices: cell/smart phones, laptops, tablet PCs, digital cameras, etc.
- Growing markets: EVs, power tools, energy storage systems for renewable energies, etc.

<sup>1</sup> Especially NiMH suffer from high self-discharge: typically 50% higher than NiCd

<sup>2</sup> Memory effect – most pronounced for NiCd



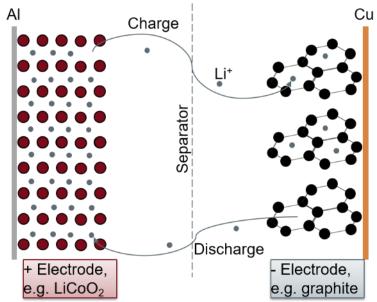
### LITHIUM ION BATTERIES - CHEMISTRY

• Graphite anode (-)

•

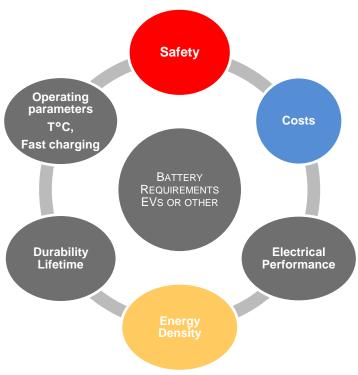
- $Li^+ + e^- + 6C \leftrightarrow LiC_6$
- Lithium Cobalt Oxide Cathode (+) (example)
  - $LiCoO_2 \leftrightarrow Li_{0.5}CoO_2 + 0.5 Li^+ + 0.5 e^-$

- Charge: Li<sup>+</sup> move from cathode to anode
  - Discharge: Li<sup>+</sup> move from anode to cathode



# LITHIUM ION TECHNOLOGY DEVELOPMENT:

- Large variety of cell technologies available on the market: NMC, NCA, LFP, spinel, LTO, etc.
- Cells are produced for different applications: high power/energy
- Depending on the geometry/type of casing different cells available: soft-case cells → pouch; hard-case cells → cylindrical and prismatic
- Cell chemistry most influencing factor regarding safety, energy density, performance and lifetime
- Other factors: electrode/cell design and manufacturing, system integration, Battery Management System





### FUTURE TRENDS

#### **General Trends**

- Replacing/reducing the amount of critical raw materials e.g. Co, graphite ⇒ Ni, Mn-rich cathode materials; Si as anode
- Applying environmentally friendly processes for materials synthesis, electrode and cell manufacturing
- Replacing Li with **multivalent ions** (e.g. Mg, Al, Ca, Zn, Ni-ion) for achieving better/higher energy storage in comparison to univalent Li- or Na-ion batteries
- Modelling and **simulating ageing** mechanisms for better estimating SoX

#### Main Aims

- Higher energy density for achieving a better driving range
- Lower costs



### **POST-LITHIUM ION CHEMISTRIES**

- Current technologies and Advanced Li-Ion
  - Ni rich NMC, high power LFP, LTO
  - High energy cell components (thin current collectors & separators, tailored electrode design)
- Tomorrow's technologies
  - **Na-ion** worse than Li-Ion, but interesting (abundance, safety, cost)
  - Thin film All-Solid-State available in small format, huge interest (safety, power density)
  - High voltage cathodes and electrolytes for Li-Ion
- Generation 2025+
  - Bulk All-Solid-State for large scale, challenge of scale up
  - **Mg-Ion** higher energy density than Li-Ion
  - Metal-air promising, but huge challenges in reversibility



### CONVENTIONAL CELL ASSEMBLY

General description of pouch cell production steps (incomplete)





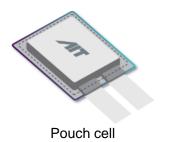
Slurry preparation



Electrode coating



Coated electrodes





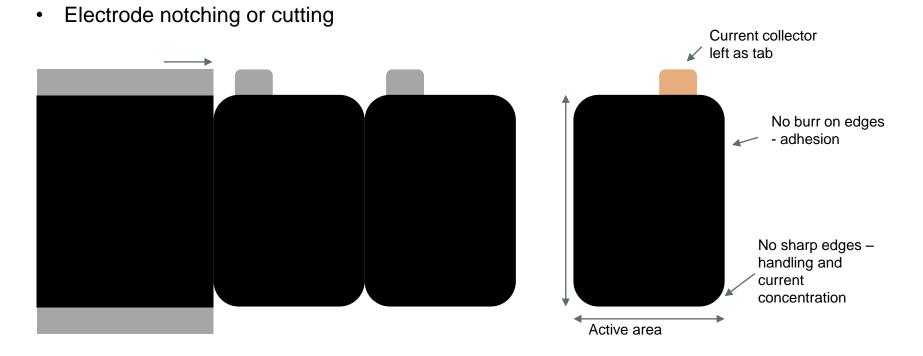
Cell assembly



calendering



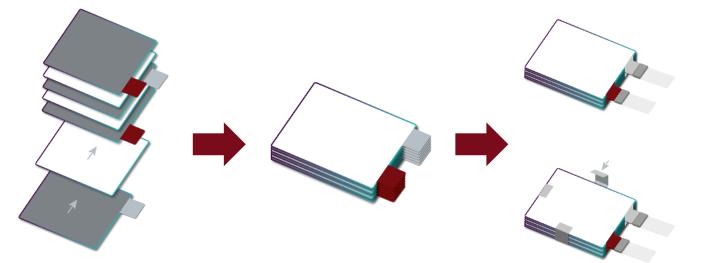
### HOW A POUCH CELL IS MADE – CUTTING





#### HOW A POUCH CELL IS MADE – STACKING

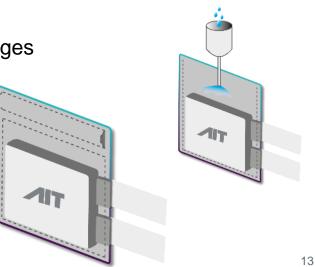
- Stacking (single stacking, z-fold, winding advantages and disadvantages)
  - Amount of electrodes determines end capacity
    - E.g. 2 mAh/cm<sup>2</sup> on a 5 x 10 cm active area:
    - $2 \text{ mAh/cm}^2 \times 50 \text{ cm}^2 = 100 \text{mAh per side} = 200 \text{ mAh per electrode (double sided)}$
    - 5 Ah pouch cell requires 25 electrode pairs (e.g. 25 cathodes, 26 anodes)





### HOW A POUCH CELL IS MADE – FILLING

- Electrolyte filling should happen under vacuum/reduced pressure
- Amount: important measure for proper operation and critical for industry
  - starting values: 4 ml/Ah for NMC; 7ml/Ah for LFP
  - Just enough, not too much (cost, gas) not too few (resistance, performance loss)
- Filling tends to be sequential total amount in 3-4 stages
  - E.g. 5Ah NMC cell = 20 ml electrolyte
  - Pull full vacuum on empty cell (remove trapped air)
  - 1/2 amount (10 ml) at 600 mbar (abs.)
  - 1/4 amount (5 ml) at 300 mbar (abs.)
  - 1/4 amount (5 ml) at 150 mbar (abs.)
  - Soaking for 30s 60s at 150 mbar (abs.)
  - Sealing at 100 mbar (abs.)





### PRIMARY FUNCTIONS OF THE BATTERY SYSTEM

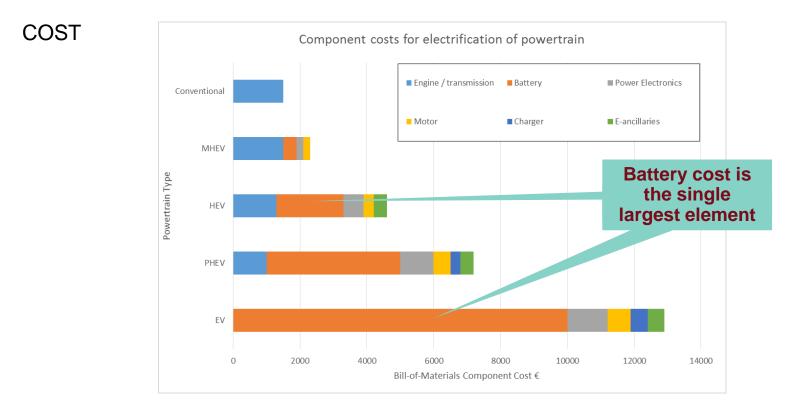
Engine starting (3kW, 2-5Wh) Conven Ancillary loads (400W average, 4kW peak, ~1kWh) tional Mild Full Hybrid Electric Vehicle

Absorb regenerated braking energy (per event) 3kW, ~50Wh for micro hybrid 13kW, ~100Wh for mild hybrid 40kW, ~ 1000Wh for HEV, PHEV, FC Support Acceleration (power and energy as above) Provide primary energy and power 10kWh – 80kWh

50-300kW



### CHALLENGE FOR COMMERCIALISATION





### CHALLENGES DIFFER BY POWERTRAIN

#### High C Rates (>20C)

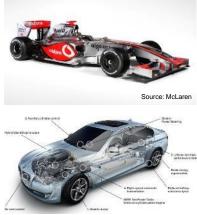
For mild and micro hybrids and high performance cars Key technical challenges are **thermal** and **impedence**, **Cost/kW** 

#### Cell Level

- Chemistry and electrode structure suited to high C
- Internal resistance of cell traded against capacity
- Thermal conductivity to cell walls/ends important
- Accurate cell level SoC understanding is critical

#### Pack level

- Liquid cooling or forced air cooling required
- BMS algorithms and sensors must respond to rapid transients active balancing sometimes required









### CHALLENGES DIFFER BY POWERTRAIN

#### Low C Rates (<5C)

For Electric Vehicles

Key technical challenges: **energy density & Cost /kWh** Although C rates rising as performance / range increases

#### Cell Level

- Capacity more important than rate of reaction
- Chemistry and electrode structure designed for durability at high depth of discharge
- Slower transients allow for simpler cell design and monitoring

#### Pack level

- Air cooling generally sufficient (unless sealed)
- Simpler BMS due to slower transients.
- Packaging volume and shape constraints









#### AUTOMOTIVE – ROBUSTNESS

- Temperatures: Hot (>50° C) and cold (<-40° C) environments
- Water on the road, shock absorption
- Safety crash impact, fire





#### AUTOMOTIVE – HIGH VOLUMES

- A typical production car makes 100,000 500,000 units/yr
- At 200 cells per pack, this is 3 cells per second or .3s/cell
- At 7000 cells per pack, this is 100 cells per second or .01s/cell





#### AUTOMOTIVE – HIGH QUALITY

- The best laptop cells have circa 1 in 200,000 failure /yr
- Laptops have typically 6-12 cells and 3 year life so premature battery failure affects <0.01%/yr</li>
- Automotive batteries have 200 to 7000 cells/car and 8-10 year life, so higher quality standards are required.

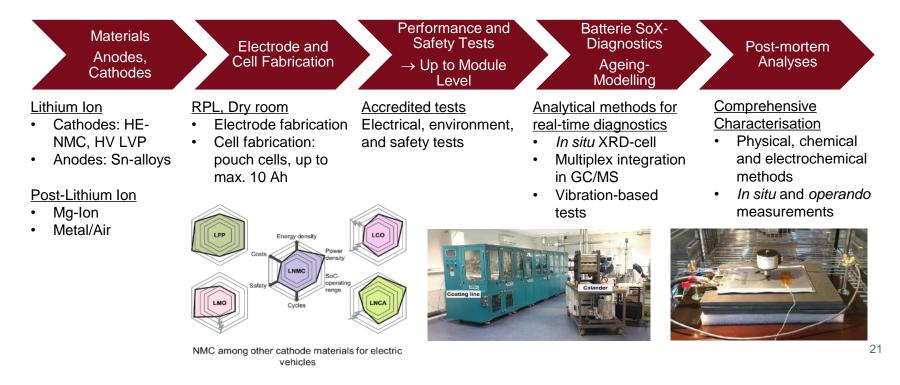






### AIT BATTERY TECHNOLOGIES: RESEARCH AREAS

#### Our aim: To cover the whole development chain







Electrolyte, Cathode and Anode Improvements for Market-near Nextgeneration Lithium Ion Batteries

#### **Project Objectives**

The objective of eCAIMAN is to bring European expertise together for developing a automotive battery cell that meets the following requirements:

- Energy density: 270 Wh/kg
- Costs: 200€/kWh
- The cells can be produced in Europe





This project is co-funded by the European Union's Horizon 2020 program under grant agreement no. 653331



### THE AIT BATTERY TECHNOLOGIES TEAM





## THANK YOU!

#### Dr. Marcus Jahn 12.11.2018

